



Chemoenzymatic Synthesis of (2S,3R)-3-Hydroxyproline from Cyclopentadiene

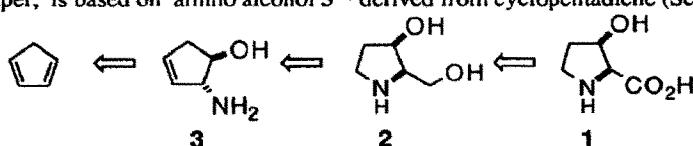
Hari Sundram, Adam Golebiowski, and Carl R. Johnson*

Department of Chemistry, Wayne State University, Detroit, Michigan 48202-3489

Abstract: Amido-alcohol **4**, derived from cyclopentadiene and resolved using *Candida antarctica* lipase B-mediated transacetylation, was transformed to (2R, 3R)-3-hydroxy-2-hydroxymethylpyrrolidine (**2**) and (2S, 3R)-3-hydroxyproline (**1**).

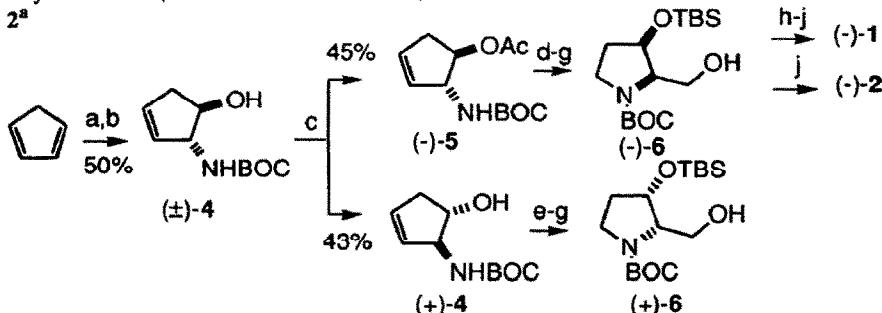
(2S, 3R)-3-Hydroxyproline (**1**) is a known, although rare, α -amino acid found in teleomycin antibiotics.^{1,2} It can be easily transformed into the Geissman-Waiss lactone³ and subsequently into a variety of pyrrolizidine alkaloids,^{3, 4} including retronecine, platynecine, and croalbinecine. (2S, 3R)-3-Hydroxyproline (**1**) and (2R, 3R)-3-hydroxy-2-hydroxymethylpyrrolidine (**2**) are present as structural units in pyrrolizidine alkaloids (e.g., australine⁵ and alexine⁶), indolizidine alkaloids (e.g., slaframine,⁷ castanospermine⁸) and unusual amino acids (e.g., detoxinine⁹). Several syntheses of *cis*-3-hydroxyproline have been reported.^{4, 10} A new approach, presented in this paper, is based on amino alcohol **3**¹¹ derived from cyclopentadiene (Scheme 1).

Scheme 1



Enzymatic resolution of racemic alcohol **4** was achieved by *Candida antarctica* lipase B (Novo Nordisk SP 435)-mediated transacetylation in isopropenyl acetate to afford acetate (-)-**5** in 45% yield (>98% ee) and alcohol (+)-**4** in 43% yield (93% ee) (Scheme 2).¹² The enantiopurity of alcohol (+)-**4** was increased to >98% by a single recrystallization (hexane : dichloromethane).

Scheme 2*



*Reagents and conditions: (a) AcOOH then NH₃/MeOH (ref.11); (b) BOC₂O, EtOAc; (c) *Candida antarctica* lipase B (Novo Nordisk SP 435), isopropenyl acetate, RT, 4 h; (d) KOH, MeOH (97%); (e) TBSCl, imidazole, DMF (98%); (f) O₃, MeOH - CH₂Cl₂ (1 : 1), then DMS; (g) NaBH₃CN, AcOH, MeOH (f + g: 71%); (h) NaOCl, TEMPO, KBr, NaHCO₃, Et₂O-H₂O; (i) NaIO₄, RuCl₃ (cat.), Me₂CO (h + i: 56%); (j) HCl, MeOH-H₂O (1: 91%; 2: 88%).

Processing of (-)-**5** by protective group exchange (O-Ac \rightarrow O-TBS),¹³ followed by ozonolysis and sodium cyanoborohydride reduction gave known prolinol (-)-**6**.^{10c, 14} Acidic hydrolysis afforded pyrrolidine **2**, while stepwise oxidation^{15, 16} (TEMPO, NaOCl then RuCl₃, NaIO₄) led, after deprotection, to 3-hydroxyproline (**1**).¹⁴ The enantiomeric 3-hydroxyprolinol derivative (+)-**6**¹⁴ was obtained from alcohol (+)-**4**.¹⁴

This chemoenzymatic method allows access to both enantiomers of 3-hydroxyproline (**1**) and the related 3-hydroxy-2-hydroxymethylpyrrolidines from cyclopentadiene. The lipase-mediated acetylation/resolution in organic media provides convenient access to enantiopure intermediates (+)- and (-)-**4** which should prove useful in the synthesis of a variety of enantiopure targets including pyrrolizidine and indolizidine alkaloids.

Acknowledgment: This work was supported by grant CHE-9223011 from the National Science Foundation. We thank Dr. S. E. Godtfredsen of Novo Nordisk A/S for the SP-435 biocatalyst.

References and notes:

- Sheehan, J.; Whitney, J. *J. Am. Chem. Soc.* **1962**, *84*, 3980.
- For a review on hydroxyprolines, see: Adams, E.; Frank, L. *Ann. Rev. Biochem.* **1980**, *49*, 1005.
- Geissman, T. A.; Waiss, A. C. *J. Org. Chem.* **1962**, *27*, 139.
- Cooper, J.; Gallagher, P. T.; Knight, D. W. *J. Chem. Soc., Chem. Commun.* **1988**, 509.
- Molyneux, R. J.; Benson, M.; Wong, R. Y.; Tropea, J. E.; Elbein, A. D. *J. Nat. Prod.* **1988**, *51*, 1198.
- Nash, R. J.; Fellows, L. E.; Dring, J. V.; Fleet, G. W. J. Derome, A. E.; Hamor, T. A.; Scofield, A. M.; and Watkin, D. J. *Tetrahedron Lett.* **1988**, *29*, 2487.
- Gardiner, R. A.; Rinehart, K. L., Jr.; Snyder, J. J.; Broquist, H. P. *J. Am. Chem. Soc.* **1968**, *90*, 5639.
- Homenschutz, L. D.; Bell, E. A.; Jewest, P. J.; Leworthy, D. P.; Pryce, R. J.; Arnold, E. and Clardy, J. *Phytochemistry* **1981**, *20*, 811.
- Kakinuma, K.; Otake, N.; Yonehara, H. *Tetrahedron Lett.* **1972**, 2509.
- (a) Hausler, J. *Liebigs Ann. Chem.* **1981**, 1073. (b) Thaning, M.; Wistrand, L.-G. *Acta Chem. Scand.* **1989**, *43*, 290. (c) Roemele, R. C.; Rapoport, H. *J. Org. Chem.* **1989**, *54*, 1866. (d) Hughes, P.; Clardy, J. *ibid* **1989**, *54*, 3260. (e) Takahata, H.; Banaba, Y.; Tajima, M. and Momose, T. *ibid* **1991**, *56*, 240. (f) Jurczak, J.; Prokopowicz, P.; Golebiowski, A. *Tetrahedron Lett.* **1993**, *34*, 7107.
- Kelly, R. C.; Schletter, I.; Stein, S. J.; Wierenga, W. *J. Am. Chem. Soc.* **1979**, *101*, 1054.
- The enzymatic acetylation of alcohol (\pm)-**4** was conducted as follows: alcohol (\pm)-**4** (500 mg, 2.6 mmol) was dissolved in isopropenyl acetate (3 mL) and to this solution SP-435 (immobilized recombinant *Candida antarctica* lipase B) (50 mg, 10 % by wt.) was added. The reaction progress was monitored by GC. At ca. 50% conversion (approx. 4.5 h) the reaction mixture was filtered, concentrated and chromatographed on silica gel (hexane : ethyl acetate; 5:1) to give acetate (-)-**5** (225 mg, 45% yield, 98% ee. by ^{19}F NMR of the Mosher ester) and alcohol (+)-**4** (260 mg, 43% yield, 93% ee. by ^{19}F NMR of the Mosher ester). For examples of other applications of the SP-435 biocatalyst from our laboratory see: Johnson, C. R.; Braun, M. P. *J. Am. Chem. Soc.* **1993**, *115*, 11103; Johnson, C. R.; Bis, S. *J. Tetrahedron Lett.* **1992**, *33*, 7287; Johnson, C. R.; Sakaguchi, H. *Synlett* **1992**, 813.
- This exchange resulted in improved yields in the subsequent TEMPO oxidation.
- Physical Data: **1** as HCl salt: mp 225 °C d.; $[\alpha]_D^{23}$ -95.5 (c 0.5, H_2O) [lit. 10b, $[\alpha]_D^{20}$ -101 (c 1.0, H_2O)]; ^1H NMR as reported.^{10f}
- 2** as HCl salt: mp 220 °C d.; $[\alpha]_D^{23}$ -12.7 (c 0.2, MeOH); ^1H & ^{13}C NMR as reported.¹⁷
- (\pm)-**4**: mp 76.0-77.0 °C; ^1H NMR (CDCl_3) δ 5.75 (dd, J = 5.7, 2.1 Hz, 1H); 5.45 (d, J = 3.9 Hz, 1H); 5.00 (bs, 1H); 4.32 (bs, 1H); 4.15 (m, 2H); 2.65 (ddd, J = 17.1, 7.2, 1.8 Hz, 1H); 2.17 (dd, J = 17.1, 2.4 Hz, 1H); 1.36 (s, 9H); ^{13}C NMR (CDCl_3) δ 156.6, 132.7, 128.2, 79.8, 79.7, 65.5, 39.4, 28.2.
- (-)-**5**: mp 71.0-72.0 °C; $[\alpha]_D^{23}$ -107.4 (c 0.7, CHCl_3); ^1H NMR (CDCl_3) δ 5.85 (m, 1H); 5.78 (m, 1H); 5.07 (m, 1H); 4.70-4.55 (m, 2H); 2.86 (dd, J = 17.7, 6.9 Hz, 1H); 2.27 (bd, J = 17.7 Hz, 1); 2.06 (s, 3H); 1.44 (s, 9H); ^{13}C NMR (CDCl_3) δ 155.1, 131.8, 129.7, 79.6, 62.0, 38.0, 28.3, 21.1.
- (+)-**4**: mp 72.0-73.0 °C; $[\alpha]_D^{23}$ +19.1 (c 1.0, CHCl_3).
- (-)-**6**: oil; $[\alpha]_D^{23}$ -35.4 (c 3.1, CHCl_3) [lit. 10c, $[\alpha]_D^{25}$ -34.4 (c 1.9, CHCl_3)].
- (+)-**6**: oil; $[\alpha]_D^{23}$ +33.5 (c 5.0, CHCl_3)
- Leanna, M. R.; Sowin, T. J.; Morton, H. E. *Tetrahedron Lett.* **1992**, *33*, 5029.
- Carlsen, P. H. J.; Katsuki, T.; Martin, V. S.; Sharpless, K. B. *J. Org. Chem.* **1981**, *46*, 3936.
- Takahata, H.; Tajima, M. Banaba, Y.; Momose, T. *Chem. Pharm. Bull.* **1989**, *37*, 2250.

(Received in USA 11 July 1994; revised 27 July 1994; accepted 28 July 1994)